A Narrative review of scientific validation of gold- and silver-based Indian medicines and their future scope

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Abstract: Metals are incinerated along with plant extracts and used as oral drugs in the Indian traditional medicines, such as Ayurveda and Siddha. Gold and silver ashes are predominantly used in cancer therapy and for treating neuronal disorders. Since the beginning of the nano-era, these ashes were investigated for their characteristics, especially the size of the particles. Numerous hypotheses were advanced to repurpose them as nanomedicines. During the last two decades, several studies were conducted studying the correlation of the particle size and the therapeutic effects. Here, we discuss the processes used to prepare gold and silver ash that result in the formation of nano- and micro-scale particles. Further, we review recent works on the ashes using modern tools and their scope, in nanomedicine. We emphasize the need to generate experimental data of high-quality to reinforce the scientific validation of these traditional medicines.

Keywords: Gold; silver; Ayurveda; Siddha; metal ashes

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Introduction

Metal-based medicines are used in the Indian medical systems, namely Ayurveda and Siddha. Gold (1), silver (2), copper (3), iron (4), and mercury (5) are some of the metals commonly used, where gold- and silver-based medicines are the most popular. The metal ashes are prepared through a special process that involves incineration of the metals that result in the formation of nano-and micro-scale particles. The metal ashes are dosed orally in the range of 1–5 mg/day for a few days along with certain adjuvants (plant or dairy products). Gold and silver ashes are used to treat neuropsychiatric disorders and tumor-related conditions. Most of these drugs were formulated around the 8th century (6) and, presently, researchers have shown great interest in studying them, using contemporary scientific tools such as electron

microscopy and X-ray diffraction instruments, etc., as well as animal and cell models. Not only did they study the properties of these ashes (7-24) efforts were also made to provide scientific interpretation behind the unique fabrication processes (8,9,16,17,20-23). Parallelly, the pharmacokinetic routes of these drugs were studied in a quantitative manner, using in vitro and in vivo models (7,10,14,25-39). As a result of these studies, the knowledge about the fabrication process and their therapeutic efficiencies has advanced. In this article, we discuss the details of the fabrication processes that are in use since the beginning and the attempts to repurpose them as nanomedicine, using the scientific data reported on the gold and silver ashes, published from 1997-2020. We present the following article in accordance with the Narrative Review reporting checklist (available at http://dx.doi.org/10.21037/ lcm-20-34).

Table 1 The plants and conditions used in the gold and silver ash preparation

Plant name and part used	ant name and part used Green synthesis of NPs		Quantity (g)/day	Levigation (days), shade drying (days)	Nos. of cow dung cakes used for incineration
Gold ash preparation (35 g of gold)				
Cajanus cajan, leaf	Gold (41)	Anti-cancer, Anti-diabetic (42)	5×35	15, 14	35
Ficus racemose, leaf	Gold (43)	Anti-cancer, Anti-diabetic (44)	4×35	12, 11	32
Ricinus communis, leaf	Gold (45)	Anti-cancer (46)	3×35	9, 8	25
Sesbania grandiflora, leaf	Zinc and iron (47)	Anti-cancer (48)	2×35	2, 1	16
Musa paradisiaca, fruit	Gold (49)	Anti-ulcer (50)	1×35	1, 1	9
Silver ash preparation (35 g of silve	er)				
Solanum nigrum, whole plant	Gold (51)	Anti-cancer, Anti-convulsion (52)	7×35	9, 8	45
Datura stramonium, whole plant	Gold (53)	Anti-cancer (54)	6×35	7,6	36
Ficus racemose, whole plant	As mentioned above	As mentioned above	5×35	5, 4	26
Chrysanthemum indicum, whole plant	Silver (55)	Anti-cancer (56)	4×35	3, 2	14

Decoding the traditional methods used in the fabrication of metal ash

The metal-medicine is prepared by an elaborate process that involves purification, levigation and incineration. It is a top-down process; the metal foils are broken down into micro-nano scale particles. Large amounts of freshly prepared plant extracts are used during the process, and more than one method is used to prepare each metal ash (40) (*Table 1*). The metal ash preparation can be broadly classified as mercury containing and formulations without mercury. A few of the methods used to prepare gold and silver ashes, respectively, and their scientific interpretation are described here.

Purification process

Purification is the first step in the fabrication of metal ashes. The metal is made into a thin foil (40–60 µm thick) and is heated to incandescence. It is then quenched in vegetable oils (sesame oil, castor oil, etc.) and other liquids (cow's urine, plant extracts, etc.). During the purification process, the metal undergoes both mechanical and chemical changes. The chemical changes mostly depend on the chemical composition of the quenchant. The metal foil is softened and sometime broken down into finer particles during the quenching (8,22,57,58). In the case of cow's urine which is rich in urea (59), quenching releases ammonia that forms ammonium hydroxide and further react with the metals to form the respective oxides (57). The metallic shine was reported to have changed to a dullish black, post the purification process of silver, possibly due to the formation of silver oxide (22,58). The cow's urine treatment has been shown to remove silica from the materials like Hematite and Magnetite which are recovered from the soil, and later used for preparing ashes (57). Similar reactions are believed to occur when the metal foils are purified in oils and plant extracts (22). Sesame oil (Sesamum indicum Linn.), is most commonly used during the metal purification process, probably because of its easy availability in India (58). Lime juice and other plant extracts are also used as quenchants (22), however, to what extent the compounds present in the quenchant contributes to the purification process is not known. In some cases, a slight weight change was observed at the end of purification (58). The purification process may eliminate certain impurities, or produce new compounds, or simply alter the microstructure and increase the brittleness. It is presumed that all of the above mechanical and chemical changes may be advantageous during the downstream processes.

Levigation and incineration process

The levigation is the second and the longest of the three processes. The purified metal is wet-grinded in a granite mortar and pestle along with plant extracts. The amount of plant extract is almost 150-200 times larger than the amount of metal used during levigation. For example, to grind 35 g of gold, almost 5 kg of plant extract was used (Table 1). In mercury containing formulations, mercury is included and the amalgam is wet-grinded. The process involves non-stop grinding over many days. The grinding during the levigation process determines the particleparticle and the particle-grinding media interactions, and it is the most common method used for reducing the particle size (60). In general, the particles may undergo amorphization, lattice deformation, agglomeration and increment in the surface area. The levigated material is collected and made into small cakes and shade dried, prior to incineration.

The dried levigated metal-cake is finally incinerated; it is placed inside a clay crucible and covered by mounting another crucible on the top. The junction is sealed, using a cotton cloth smeared with wet clay, limiting the supply of air inside the crucible. The cow dung cakes are used as heating fuel where, a maximum of 800-900 °C is reached (8,22,58,61). However, the temperature inside the crucible is expected to be much higher. In the recent years, electrical furnace is substituted for cow dung-based incineration (58). It is a general rule that the levigationincineration cycle is repeated multiple times, in some cases, even 1,000 times until the right texture and color are achieved (16). The objective of the repeated cycles of levigation and incineration processes is, not only to achieve a small particle size, but also to complete certain chemical processes, in this case, the oxidation of the metals (16). In most cases, the end product is the oxide of the corresponding metal and, in cases where sulphur is included, the sulphide of the metals is also present in the metal ash (8,16,19,20,25).

In gold ash fabrication using plant extracts, (*Table 1*), 35 g of gold is wet-grinded in 175 g of *Cajanus cajan*, leaf juice, each day for a period of 15 days after which the metalcake is dried for 14 days and incinerated. Followed by *Ficus racemose*, leaf juice (140 g, 12 days), *Ricinus communis*, leaf juice (105 g, 9 days), *Sesbania grandiflora*, leaf juice (70 g, 2 days) and finally in 35 g of banana juice for one day. Similarly, for silver ash preparation, 35 g of silver is wetgrinded in 245 g of *Solanum nigrum*, whole plant juice, each day for a period of 9 days; the cake is dried for 8 days and incinerated. Followed by *Datura stramonium*, (210 g, 7 days), *Ficus racemose* (175 g, 5 days), and finally in *Chrysanthemum indicum* (140 g, 3 days). Incineration is performed after each levigation process using the prescribed amount of heat (*Table 1*).

The gold ash preparation using mercury was explained in a previous study (8). In brief, purified gold were added to mercury (1:2, ratio), and grinded into amalgam. The amalgam was placed into a clay crucible along with sulphur in a 1:1 ratio and incinerated (~900 °C). The grinding and incineration processes were repeated 42 times. The silver ash preparation method with mercury, used in our previous work (10), is described below. A 4 kg, silver foil, was amalgamated in 1 kg of mercury, and levigated with the Pergularia Daemiar leaf juice, 10 kg. The end product is made into small cakes, dried and incinerated at around 700 °C and the grinding and incineration cycles are repeated two more times. Finally, egg shells 0.5 kg, are added to the incinerated silver and levigated again with the Daemic leaf juice and incinerated repeatedly until a fine powder is obtained.

It is believed that the plants were chosen based on their local and seasonal availability and medicinal properties (*Table 1*). The antioxidants and the polyols present in the plant extracts act as both reducing and stabilizing agents (62). Most of these plants possess anti-cancerous activity among other notable therapeutic properties (*Table 1*). It has to be mentioned that the plants used in this fabrication process are today used in the green synthesis of metal nanoparticles (NPs), *Table 1*. Whether the medicinal properties of the plants are transferred to the final product is controversial because of the high temperatures at which they are incinerated.

The fate of the plant-based components present in the incinerated material is important in order to decide their therapeutic contribution in the final product. The incineration temperature determines the composition of the plant ashes (20). When combusted at temperatures below 450 °C, the main component of the incineration of the plant-based materials will be carbon. As the temperature increases above 450 °C, the carbon becomes volatile and the ash mostly comprises of inorganic carbonates, containing elements such as calcium, magnesium, silicon, sodium, potassium, etc. Above 580 °C, most of the minerals are oxidized, releasing carbon dioxide (63). However, the limited supply of oxygen, might render the carbon compounds incompletely oxidized, to form carbon monoxide. It has to be noted that carbon monoxide is a reducing agent that might have an effect on the oxidation of the metal itself (64). A well finished ash is believed to be devoid of metal remnants or carbonaceous materials (16). Some studies have shown that postincineration metal ash is devoid of any organic compounds (8,17,21), while others have reported the presence of a carbonaceous material (19,65,66). However, it is suggested that the plants were simply used for their reducing properties and for mechanically easing the grinding process, rather than for their medicinal values.

Further, we think that the inclusion of large amount of plant extract could result in an even distribution of heat during the incineration. In addition, the presence of large amounts of carbon during the repeated incineration processes may have an additional advantage, lowering the vaporization temperature of the metal components. A similar effect of carbon has been found for the gold recovery form sewage sludge. In the absence of carbon, gold is usually released at a temperature around 600 °C and vaporization occurs at 1,000 °C. In the presence of carbon, the gold is released to gas phase at a lower temperature, around 400 °C (67). This temperature is significant, especially when metals such as mercury are present. Given that during the final incineration, the amount of plant components would be much higher when compared with that of the metal itself, the fate of the organic compounds must be probed in those preparations. The combusted plant product contains high amounts of alkaline earth metal oxides that give the trademark whitish color of the ash. The color of pure gold oxide (Au_2O_3) and silver oxide (Ag_2O) is reddish brown and black, respectively (16). The gold and silver ashes (prepared without sulphur) are white powders with a yellow and black tint, respectively.

The metal ashes prepared using mercury are considered superior to the ones that only use plant extracts, probably because the bulk gold or silver will be dispersed into small particles in the mercury matrix, which during incineration results in finer particles (6). The ductility of gold was also observed to be reduced and gold often crystalizes when gold amalgam is subjected to high temperatures (68). The absence of mercury in the ashes were considered as an indication of a high quality incineration (8,20,21) and long incineration times have proved to be beneficial to remove mercury from the ash (8). However, the same cannot be said about all the preparations, a notable amount of mercury is present in many of the metal ashes (17,19,20,24).

Particle size and chemical composition of the meal ashes

Following the recent development of electron microscopy and X-ray diffraction techniques, the metal ashes were studied for their particle size and chemical composition. Transmission electron microscope (TEM) (8) and scanning electron microscopy (SEM) (11,23) image the morphological appearance of the particles. Dynamic/static light scattering (DLS/SLS), provides the size distribution of the particles. In the case of non-spherical particles such as metal ash, the hydrodynamic radii are analyzed, based on the equivalent sphere (69). Determination of the elementalchemical composition of the particles was done, using the following methods: Energy Dispersive Spectroscopy (EDS/EDX), X-ray fluorescence (XRF) spectrometry, Inductively coupled plasma mass spectrometry (ICP-MS) and Atomic Absorption Spectroscopy (AAS). The crystalline nature and the crystallite size are determined using X-ray Powder Diffraction (XRD). In the EDX and XRF analysis, the sample is bombarded by electron beam and X-ray, respectively; the X-ray energy emitted is used to identify the elements present. In the ICP-MS, the acid digested metal ashes are plasma ionized, then passed through a quadrupole mass analyser, where they are separated and detected according to their mass-charge ratio (70). In the case of, AAS, the absorption spectrum of the ionized atom is used to detect the elements present in the sample (71). The metal ash is highly heterogeneous in terms of elemental composition and the analysis method plays a huge role in the chemical characterization. Trace elements (concentrations below 0.01 wt%) and the lighter elements (<Na), cannot be detected by EDX. In addition, it has a limited accuracy because the energy peaks overlap among different elements. On the other hand, the irradiation area in XRF is larger than in EDX, and results in better sensitivity when analyzing heterogeneous samples. AAS is reported to quantify elements with better accuracy than the EDX (71), while ICP-MS is more precise than AAS (72). The XRD is used to calculate the crystallite size, and a complimentary method to analyses the chemical composition (13,23). Irrespective of the advantages and limitations presented for the above methods, the studies allowed a closer look at the metal ash and into the fabrication process itself.

In 2007, gold ash was imaged for the first time, by using (TEM) and globular NPs with an average size of 56–57 nm were found (8). In other publications, a mixture of nano and micro size particles was found, *Table 2*. The NPs were

	In-house (8)	IMPCOPS (13)	AVK (21)	AVK (21) In-house (23)	In-house (24)	JIMP (11)	JIMP (11) In-house (9)
Particle size	56–57 nm (TEM)	I	~10 µm (SEM)	~10 µm (SEM) 1-10 µm (SEM)	<9 µm (SEM)	5 µm (SEM)	5 µm (SEM) ~5 µm (FESEM)
Crystallite size	23–37 nm (XRD)	28–35 nm (XRD)	I	I	I	I	I
Hydrodynamic diameter	I	S1-717 nm; S2- 669 nm (DLS)	I	I	S1-7.55 S2-8.44,	I	I
					S3-9.97 µm (SLS)		

shown to aggregate into particles as large as 5 µm (11). Similar studies showed silver ash to be made up of, nano and micro particles (10,12,15,17-20) (*Table 3*). The gold content in ash ranged from 50% to 98%, along with varying amounts of other elements (7,8,11,14,15,21,24). The amount of silver also varied between reports 52–81.77% (15,17-19,22). The presence of other elements in the ashes, such as Fe, Zn, Mn, Cu, Ca, etc., was also reported (*Tables 4,5*). The trace elements are considered a source of the micronutrients required for the body (73).

The amount silver was found to be reduced with increased number of incineration (22), Table 5. It was reported that after 42 cycles of incineration, the particle size of gold ash was found to be around 56-57 nm (8), nonetheless the particle size in case of lesser or greater number of incineration was not reported. It is known that repeated incinerations of metal particles may result in their aggregation (74). In most methods, the prescribed amount of incineration as shown in the Table 1, is gradually reduced over the consecutive steps. However, there are other methods where the same amount of heat is used throughout the incineration cycles (8). Therefore, the effect of repeated cycles of levigation and incineration is not known. The difference in the size and composition is mostly because of the different fabrication processes (17). In addition, the method used to analyze the metal ash may also have caused the difference in the size and elemental composition (17).

Pre-clinical and clinical analysis of metal ashes

The journey of metal ash inside the body

Unlike the modern nanomedicine, the metal ashes were not designed to target a specific cell type or a receptor. The indications are specified in terms of a particular pathological symptom and, in few cases, a particular organ (40). The pathophysiology belongs to the medieval period, therefore, the drug route, or the pharmacokinetic activity does not coincide with the current understanding. Hence, we can assume that the metal ashes would follow the same fate as any foreign particle that enters the systemic circulation.

In general, before particle enters the systemic circulation, the particle has to pass through the orogastrointestinal tract. The conditions inside the digestive tract are not uniform, the cell type, thickness of the mucus lining, pH, all vary throughout the tract. Immediately after entering the body, the metal oxide NPs are coated by the protein and amino acids (protein corona) in the body fluids. It is

Pa			inite alumeter, reported by furio		
	In-house (12)	In-house (18)	SBAB (19)	In-house (20)	IMPCOPS (10)
Particle size	16 nm (NA)	10–60 nm (SEM)	500 nm to a few µm (SEM)	1.04 µm (SLS)	300 nm–1.7 μm (SEM)
Crystallite size	-	-	_	-	54.6 nm (XRD)
Hydrodynamic diameter	-	-	-	-	530 nm–1.1 μm (DLS)

Table 3 The silver ash's particle size, crystallite size and hydrodynamic diameter, reported by various studies

SBAB, Shree Badyanathan Ayurved Bhawan; NA, not available.

reported that the protein corona is formed within 5 min of exposure to the proteins (75). Once consumed, the particle briefly encounters the mucus layer (thickness, 70–100 µm) in the oral cavity and the esophagus and reaches the stomach filled with acidic digestive juices (thickness, 897-1,354 µm) (76). In general, the positively charged, hydrophilic particles and the Zwitterionic particles, have a high chance of penetrating through the mucus layer. On reaching the cell's surface, in most cases the polarized enterocytes, the particles sterically interact with the cell membrane to gain access into the cell through the transcellular routes. The larger ones, with the particle size >1 µm are taken up by micropinocytosis, ~200 nm by the clathrin- and ~80 nm by caveolinmediated endocytosis (76). Smaller particle in the size <20 nm may cross the intestinal barrier through the paracellular routes to finally gain access into the systemic circulation. Depending upon, whether the NP is functionalized with specific protein receptors [e.g., Transferrin, g-protein, etc. (77)] to target specific sites, they will circulate all over the body and accumulate in various organs such as liver, spleen, lungs, brain, kidney, testicles, ovaries and skin. The liver is the major clearance routes of the metal oxides >5 nm. They are secreted through bile and are excreted from the body (78). While kidney clears out metal oxides <5 nm through the urine (78), recently it was found that spleen also clears out metal particles from the body.

The analgesic property of gold and silver was for the first time investigated in 1997 using animal models (32-34). Later, numerous studies reported gold ash's effect on the immune-stimulant activity in brain macrophages (35), as anxiolytic, anti-depressant, anti-cataleptic in behaviorally despaired animals (14,38), free radical scavenging activity in animal models (7), and antioxidant activity in ischemia animal models (36). Similarly, for the silver ash, the sedative-hypnotic activity (26), antibacterial activity (18), anti-inflammatory activity (25), analgesic activity (12), and hypolipidimic activity (37), were reported. The effect of silver ash on brain diseases was also studied (28,30).

Silver ash particles were found to be accumulated in the skin, spleen (malpighian body), liver (histiocytes), kidneys (renal glomerulus), and lymph nodes of the animal models (12). From the above studies, it is clear that the metal ashes were able to cross various barriers and reach the organs. However, the duration of the retention in several organs, the mechanism through which they are excreted from the body is not completely known.

Intercellular fate of the metal ashes

The understanding of how metallic particles behave once they enter the cells is evolving. Most of the particles that enter the cell are processed via. the endosomal-lysosomal system. During the endocytosis process, the particles are encapsulated in a structure called endosome, which gradually transforms into lysosomes. Lysosomes have a highly acidic (pH upto, 4.5) microenvironment that disintegrate the particles and release them into the cell or dispatch them out of the cell via exocytosis (79). Any internalized particle may have the following effect on the cells: they interact with the cell components and induce oxidative stress, disrupt the cytoskeleton, lower the cell proliferation rate, hinder the cell differentiation and on entering the cell nucleus, they may damage the DNA (80). All the above events could eventually lead to the cell death and are exploited for treating diseases like cancer. There are only a few studies exploring the intracellular processing of metal ash particles. Recently, a study on the gold ash cellular uptake confirmed that the particles entered the cells via clathrin-dependent receptor-mediated endocytosis (11). The metal ash particles were accumulated near the perinucleus region (10,11). In another study, the uptake and localization of the silver ash particles were found to vary based on the cell types (10). A long term (1 year) retention in vitro study showed that, gold NPs and iron oxide coated gold NPs, were found to disintegrate and recrystallize inside the cells (79,81). The metal ash particles could also follow

Element %	JIMP (11)	In-house (7)	In-house (8)	In-house (21)	_	In-house (24)		In-house (9)	IMPCC	IMPCOPS (13)	ARS (14)
(Method)	ICP-MS	AAS	AAS	AAS	AAS, S 1	AAS, S 2	AAS, S 3	EDAX	EDAX, S 1	EDAX, S 2	NA
Au	56.88	20.34	92	>90	98.00	98.33	98.41	60.98	88.10	91.2	46.9
As	0.15	0.17	I	I	I	I	I	26.87	9.95	9.56	I
Fe	0.29	3.90	I	I	0.17	0.10	0.12	1.41	I	I	I
0	I	I	I	I	I	I	I	8.83	I	I	I
Ag	I	I	I	0.05	0.05	0.04	0.04	I	I	I	I
Pb	I	0.03	I	I	<0.005	<0.005	<0.005	I	I	I	I
Cu	I	0.18	I	I	0.021	0.020	0.013	I	I	I	I
Hg	I	I	BDL	I	<0.001	<0.001	<0.001	I	I	I	0.15 ppm
Zn	I	74 ppm	I	I	0.020	0.019	0.021	0.26	I	I	I
Nb	I	I	I	I	I	I	I	I	0.56	0.72	I
Mg	1.8	2.08	I	I	I	I	I	I	I	I	I
Са	1.4	1.96	I	I	I	I	I	I	I	I	I
ЧЛ	0.037	I	I	I	I	I	I	I	I	I	I
Si	0.29	I	I	I	I	I	I	1.64	I	I	I
Ba	I	0.33	I	I	I	I	I	I	I	I	I
Sr	I	1.28	I	I	I	I	I	I	I	I	I
AI	I	8.60	I	I	I	I	I	I	I	I	I
Co	I	70 ppm	I	I	I	I	I	I	I	I	I
ïz	0.02	75 ppm	I	I	I	I	I	I	I	I	I

ARS, Ayurveda Ras Shala, Pune.

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Table 5 The list of elements pr	resents in silver a	ish reported in	various studies

Element %	SB	AB (19)	In-house (18)	IMPCOPS (17) mg/g	IMPCOPS (17)	In-hou	se (20)	In-house (15)	In-hou	ise (22)
(Method)	EDX	ICP-AES	ICP-AES	ICP-OES	EDX	XRF	EDS	NA	NA, S 1	NA, S 2
Ag	48.34	65.84	70.56	325	6.91	84.09	49.50	52–59	81	80.82
S	13.99	14.41	-	-	-	7.92	31.64	0.675	13.15	14.32
С	3.79	1.69	-	-	30.75	-	-	-	_	-
N	-	0.28	-	-	-	-	-	-	-	-
0	22.49	-	-	-	46.49	-	-	-	-	-
Н	-	0.56	-	-	-	-	-	-	-	-
As	2.31	-	-	0.0195	-	-	-	-	-	-
Fe	-	-	-	1.4800	-	0.61	0.56	-	-	-
Pb	-	-	-	0.0351	-	-	-	-	-	-
Cu	-	-	-	0.2530	-	1.04	1.87	-	-	-
Hg	-	1458 ppm	-	1.3220	-	-	-	-	-	-
Zn	-	_	-	0.1575	-	-	-	-	_	-
Mg	-	-	-	2.2120	1.27	0.07	3.15	-	-	-
Ca	-	-	-	98.0530	-	1.37	2.25	10.769	-	-
Mn	-	-	-	0.0338	-	-	-	-	-	-
Si	-	-	-	-	0.75	0.38	2.90	-	-	-
Al	0.52	-	-	BDL	0.48	0.10	0.88	-	-	-
Ni	-	-	-	0.0130	-	-	-	-	-	-
Cd	-	-	-	BDL	-	-	-	-	-	-
Cr	-	-	-	0.0310	-	-	-	-	-	-
CI	-	-	-	-	-	0.29	1.53	-	-	-
К	-	-	-	-	-	3.31	5.62	-	-	-
Р	-	-	-	-	-	0.25	-	-	-	-
Rh	-	-	-	-	-	0.32	0.10	-	-	-
Os	-	-	-	-	_	0.02	-	-	_	_

SBAB, Shree Baidyanath Ayurved Bhawan.

a similar fate when they are retained in the cells for longer periods.

The metal ashes were considered safe, on the bases that they cannot be metabolized in the body (82), which is not the case. The metal ions are reported to affect various systems- immune (83), nervous (84) and reproductive (85). They release metal ions that generate reactive oxygen species (ROS) in *in vivo* and *in vitro* conditions, which plays a key role in the cell survival (86). The ROS cause oxidative stress and affects the cell functions in numerous ways; it interferes with cell-signaling mechanisms and also damages the biomolecules. The DNA damage and inhibition of DNA repair are some of the events that can transform a normal cell into a cancer cell (87). There is a preferential uptake of gold NPs by the cancer cells when compared with normal cells (88). The gold NPs have also shown to cause both necrosis and apoptosis in cancer cells, while only apoptosis in the normal cells. It is not clear if a similar

preference is shown by the ash particles and if it is non-toxic for the healthy cells.

Gold ash is reported for their anti-oxidant, free radical scavenging properties (9,36). Another study on gold ash also showed significantly increased superoxide dismutase and catalase activity, two enzymes that reduce free radical concentrations in the body (7). Further, the gold ash was found to be effective in treating various solid malignancies, in particular colon cancer (39). Out of the 39 patients, 17 showed complete (disappearance of all target lesions) or partial response (at least a 30% decrease in the sum of diameters of target lesions), and no significant side effects were observed for the entire duration of 1 years when 50 mg/kg/day was administered. At cell level, gold ash was found to exhibit anti-tumor activity, by disrupting the cytoskeletons of the tumor cells (11).

Toxicity studies on metal ashes

So far, cell-level toxicity studies using MTT assay show that the gold ash is not toxic to both cancer [L929 fibroblast cells (13) and HeLa cells) (11) and non-cancer cells (HFF-1 cells (11)]. Animal study on gold ash have also showed that they do not cause any adverse effect (27). However the gold ash is speculated to have caused adverse effect in a jaundice patient (31). Further, the presence of mercury, arsenic and lead in the gold ashes has also raised concern about their toxic impact in the body (89).

Silver ash was found to lower the lipid levels in the blood, liver and muscles (37), probably due to the impaired lipid metabolism caused by the accumulation of silver ash particles in the liver. In another study, silver ash particles were found aggregated in various organs and a bluish black skin pigmentation was observed in some of the animals (12). These observations highlight the pitfalls in using the silver ashes. Further, the possibility that the gold and silver ash particles could disrupt the tight junctions (13,19) and travel across tight barriers such as BBB is expected to have implication on the brain tissues. It is thought that the adjuvants could prevent the adverse side effects during the administration of metal ashes (90), however there is no study demonstrating the above.

More than one disease can be treated using the same metal by simply varying the adjuvants (91). For example, for shivering, the gold ash is administered along with betel leaf juice, for loss of weight, and discharge of glucose in urine and for rejuvenating effect it is consumed along with honey and butter, respectively (40). There are two hypotheses based on the administration of metal ashes along with adjuvants. The first suggests that, the adjuvants enhance the therapeutic efficiency of the metal ashes and according to the second one; the metal ash increases the therapeutic effect of the phytochemicals present in the adjuvant. A recent study demonstrated that, when gold ash was administered along with pepper (Piper nigrum) and ghee (29), the uptake of the gold ash was enhanced, confirmed by measuring the gold levels in the blood. In order to verify other hypothesis, cell permeability assay on CaCo-2 cell monolayers was performed for the silver ashes. It was found that the silver ash facilitated paracellular transport of AlexaFluor dye molecules, 884.91 g/mol, across the monolayer (19). This study raised the possibility that the metal ashes could have been used to increase the uptake of the phytochemicals present in the plant-based adjuvants. However, in some cases, the metal ashes were consumed simply along warm water and dairy products, which do not have specific pharmacological effect.

Nano-dimension of the metal in the ashes

In the context of the contemporary sciences, the incinerated ashes used in the Indian medicines were repurposed as nanomedicines (6,82,92). Nanomedicines have at least one of their dimensions below 100 nm for \geq 50% of the particles (93), or have surface to volume ratios exceeding 60 m^2/cm^3 or exhibit a physical, chemical and biological effect that attributes to its size, even if it falls outside the nano-range, up to 1,000 nm (94). According to these definitions, the metal ashes are classified as nanomedicine (17). However, not all metal ash particles are <100 nm or <1,000 nm, the percentage of NPs in the metal ash is not uniform (17,95) (Tables 2,3). Gold ash was reported to have only 10% of the particles below 1 µm (24) and for silver ash only <10% of the particles were below 188.7 nm (17). In addition to this, the nano-scale phenomenon are not reported for the metal ashes (2). Probably, due the heterogeneous size distribution or the presence other elements in the metal ashes could mask the nano-size effect. Therefore, it is for reconsideration if the metal ashes can be generalized as nanomedicine.

In a biocompatibility study, unlike metal/metal oxides (75), the gold ash particle did not adsorb proteins, even after 1 h, incubation (13), probably because of the non-uniform shape of the gold ash particles (96). Nonetheless, the therapeutic property of the metal

ashes can mainly be attributed to the particle size that favors their cellular uptake (36,97) and their ability to generate ionic species (98). Inspired from Ayurveda, in a recent study, a metal ash-like formulation comprising NPs synthesized through green synthesis and a cocktail of herbal extracts, was reported to be effective on breast cancer patients (89). It is named "nano *swarna bashma*", however it is completely different from the gold ash used in Ayurveda, except for the fact that gold is used in both the preparations.

Considering that the major constituents in the metal ashes are the respective metal oxides, the metal ash particles may share similar interactions with the cells and organs as that of the artificially synthesized metal oxide NPs. However, the heterogeneity in size and shape and the presence of an undesirable amount of elements such as mercury, arsenic and lead in the metal ashes may have different therapeutic and toxic implications (99).

Conclusions

In this review, we have summarized the fabrication and therapeutic properties of two important metal-based medicines used in the traditional Indian medicine - gold and silver bhasmas/parpams, also called ashes. The process of fabrication was described, following the ancient texts and methods that are still in use today. However, during the last decades, by using the modern physical methods, the lack of uniformity in terms of composition has been demonstrated for products fabricated by various companies. This is, probably, one of the reasons for the often-contradictory results, regarding the particle-cell interactions. Hence, the scientific validation of these medicines is not conclusive (100); this is a process to be continued and improved once the metal ashes will become more uniform and the analytical methods will progress in terms of accuracy. High quality SEM and TEM measurements of metal-based medicines are required in order to ascertain (or contradict) the assumption, regarding the role of the nano-sized particles in the therapeutic properties of metal-based Ayurvedic medicines. For now, it is difficult to accept the idea that particles incinerated several times at high temperatures would be nano-sized.

There is a tendency among the Indian medicine practitioners to advocate for the safety of these metallic preparations, argue that they are natural products and do not cause any adverse effect (31), or simply because it is an ancient medicine and in practice for many years (101,102). It cannot be denied that the metal ashes were formulated for their safe use and are time tested. The metal ash usage comes with detailed cautionary instructions, about their dosage, adjuvants to be used, anti-dotes in case of adverse effects, dietary advice, etc. (40). However, there is a huge knowledge gap in the understanding their behavior at the cell-level, organ-level and in the entire body. More studies on the sub-cellular behaviors of metal ash particles are required to precisely elucidate their impact at different levels.

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Footnote

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