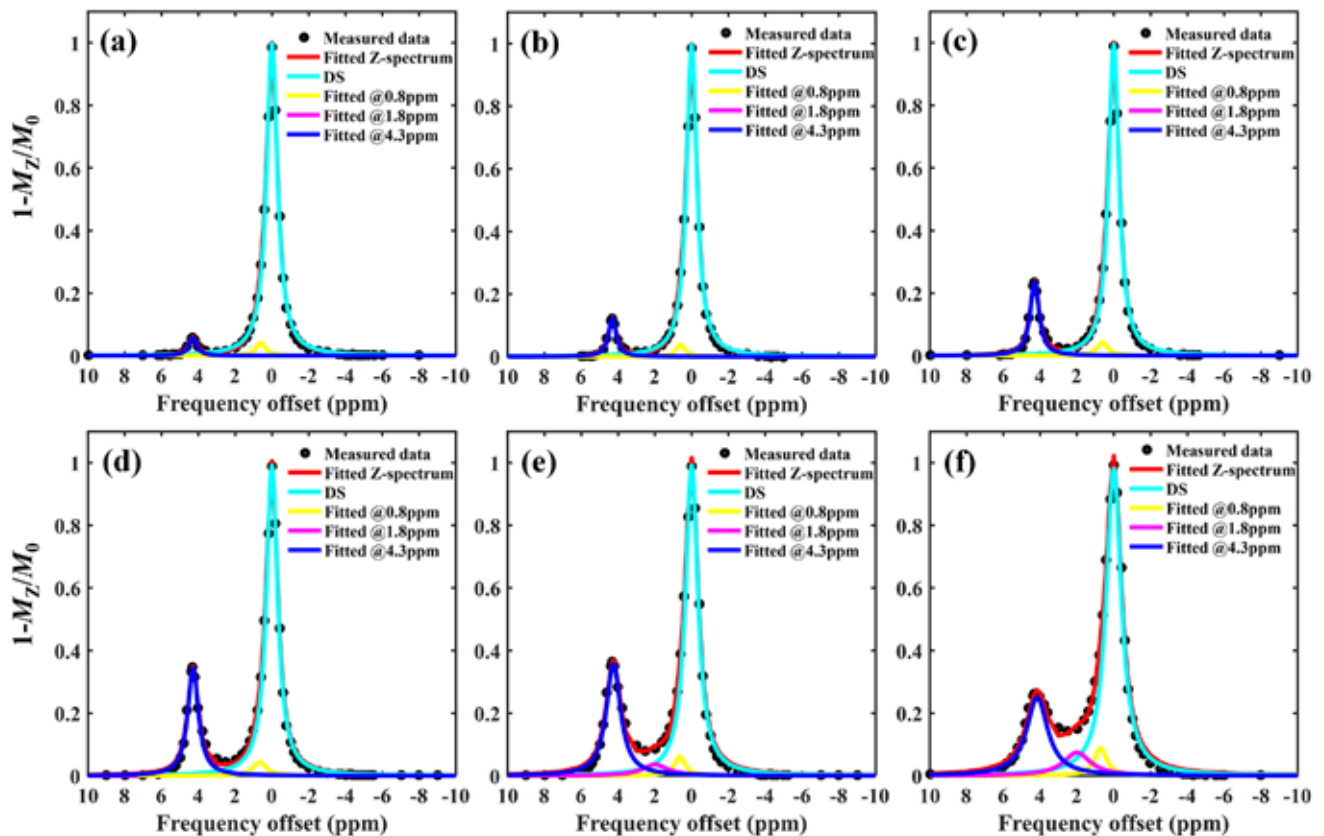


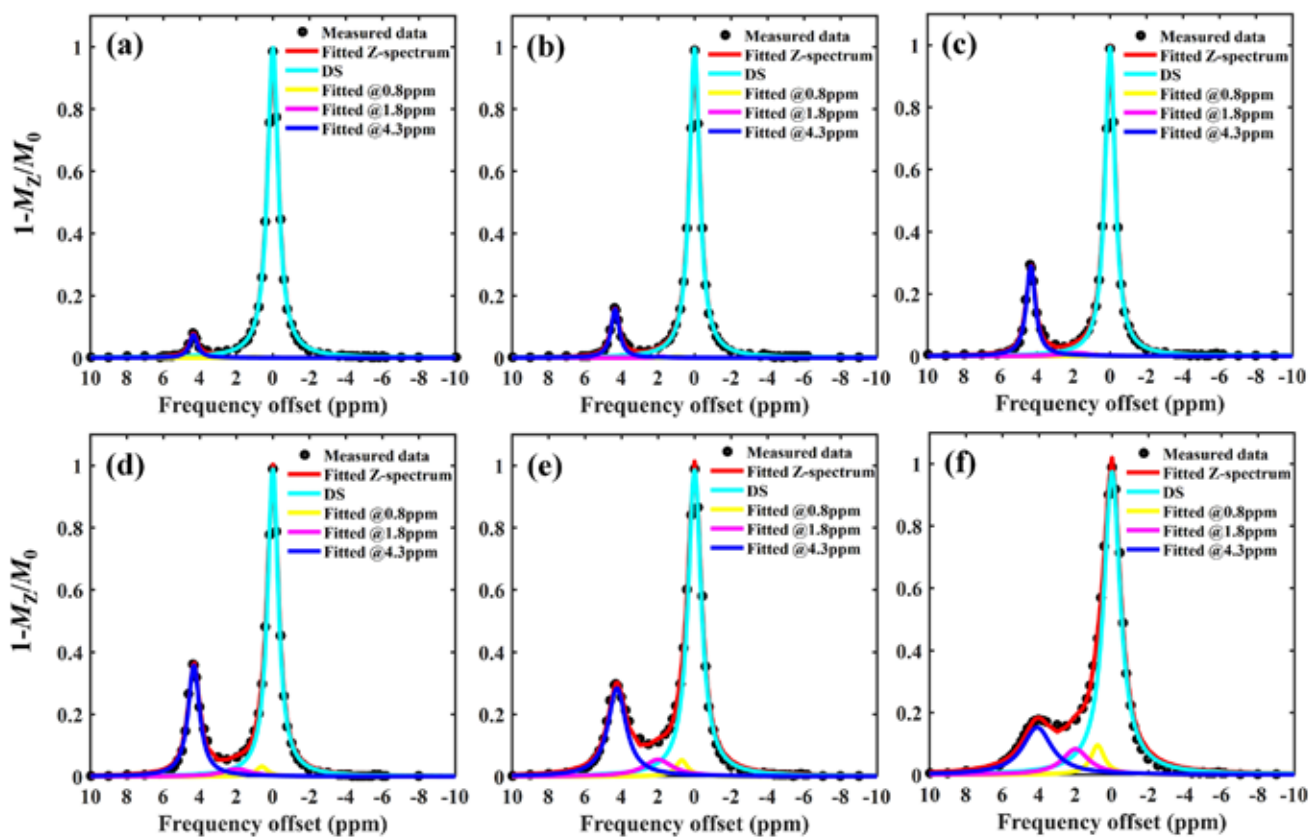
## Methods

### CT experiments

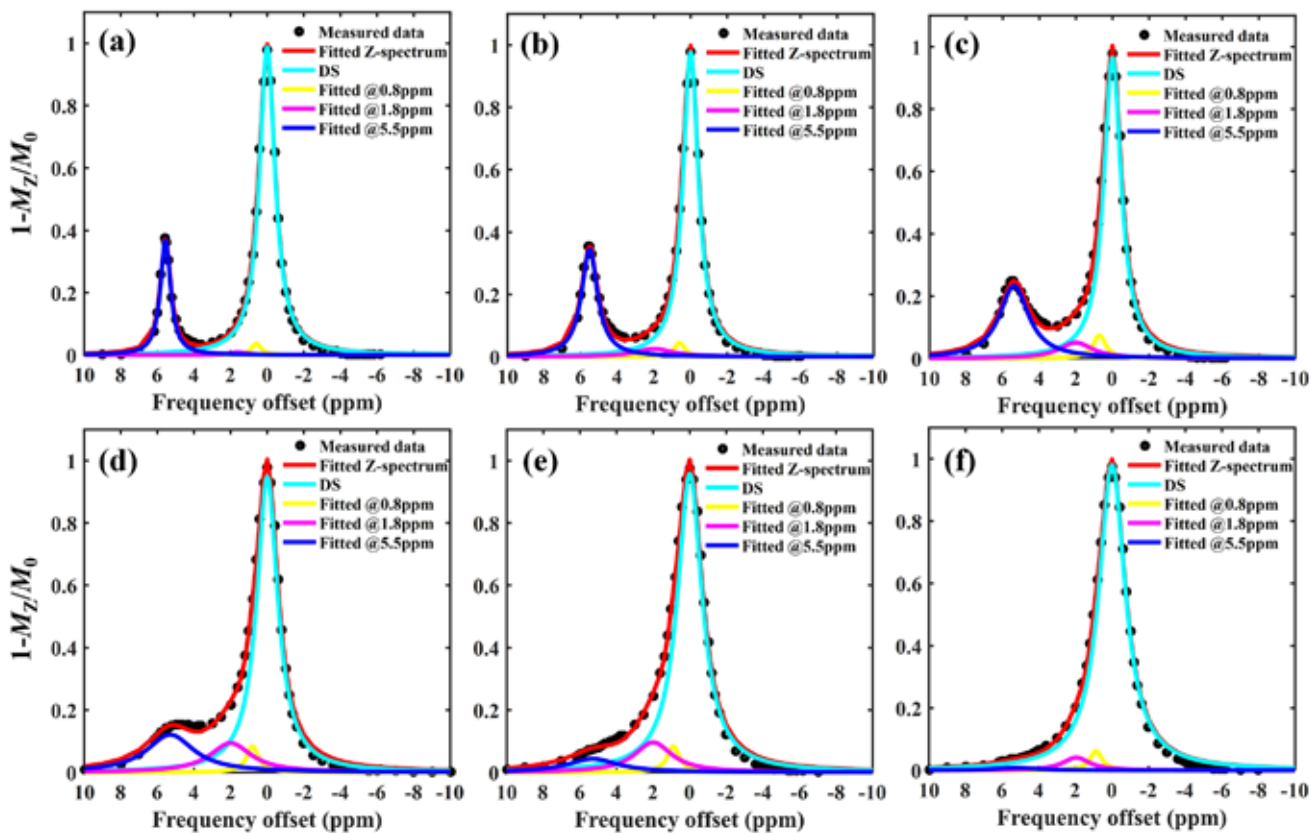
The imaging protocol was performed on a micro-CT scanner (Hitachi-Aloka, Tokyo, Japan). The phantoms of iodixanol and iobitridol with the same iodine concentration (1.25–40 mM) were scanned for the measurement of their concentrations in the kidneys. The rats (n=8) were anesthetized by 3% pentobarbital sodium through intraperitoneal injection at a dose of 30 mg/kg body weight. After all the animals were in a fully anesthetized state, iodixanol was firstly injected via a catheter into the tail vein, with the injection procedure the same as that for the *in vivo* MRI experiment. The body temperature was maintained by a hot water bag and supervised by anal temperature detection. Dynamic CT images were acquired during the period of post-injection to 24 minutes, with the following parameters: 1,024 projections, 50 kV, 150 mA, 10 seconds exposure time, FOV =81.5 mm. The total scanning time was approximately 4 minutes. Reconstructed CT images were analyzed using MATLAB, and the CT values (Hounsfield units; HU) were measured in a manually defined kidney ROI. A calibration curve (CT value versus iodine concentration) was derived at 50 kV using phantoms filled with 7 different iodine concentrations ranging from 0.625 to 40 mM. After subtraction from the pre-injection image,  $\Delta$ HU were calculated to determine the quantitative iodine concentrations. Afterwards, iodixanol was filtrated almost entirely from the kidney at an interval of 2 hours, iobitridol was then injected and the above experiment repeated. Finally, the obtained ratio of iodine concentration was converted into the ratio of exchangeable protons at 4.3 and 5.5 ppm.



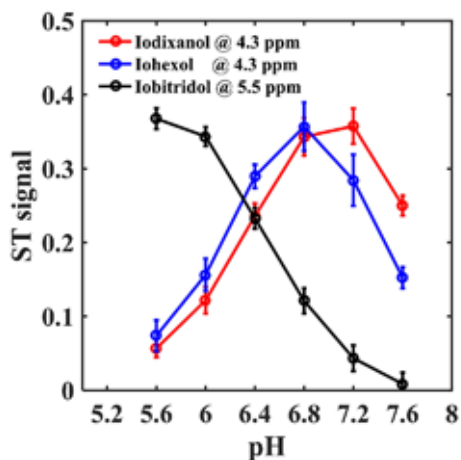
**Figure S1** *In vitro* Z-spectra of iodixanol phantom and their fitting results using a four-pool Lorentzian model under the experimental conditions of saturation power =1.5  $\mu$ T, saturation time =5 s, temperature =37  $^{\circ}$ C, agent concentration =30 mM, and different pH values: (A) pH=5.6, (B) pH=6.0, (C) pH=6.4, (D) pH=6.8, (E) pH=7.2, and (F) pH=7.6. DS, direct water saturation.



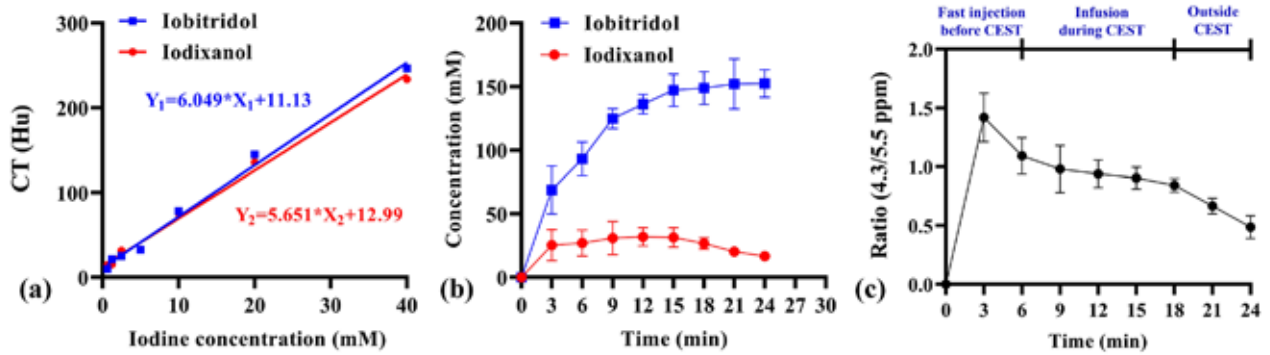
**Figure S2** *In vitro* Z-spectra of iohexol phantom and their fitting results using a four-pool Lorentzian model under the experimental conditions of saturation power =1.5  $\mu$ T, saturation time =5 s, temperature =37  $^{\circ}$ C, agent concentration =60 mM, and different pH values: (A) pH=5.6, (B) pH=6.0, (C) pH=6.4, (D) pH=6.8, (E) pH=7.2, (F) pH=7.6. DS, direct water saturation.



**Figure S3** *In vitro* Z-spectra of iobitridol phantom and their fitting results using a four-pool Lorentzian model under the experimental conditions of saturation power =1.5  $\mu$ T, saturation time =5 s, temperature =7  $^{\circ}$ C, agent concentration =120 mM, and different pH values: (A) pH=5.6, (B) pH=6.0, (C) pH=6.4, (D) pH=6.8, (E) pH=7.2, (F) pH=7.6. DS, direct water saturation.



**Figure S4** Quantified ST signals from amide protons of iodixanol and iohexol located at 4.3 ppm and iobitridol at 5.5 ppm, respectively. ST, saturation transfer.



**Figure S5** The ratio variation of two nonequivalent amide protons at 4.3 ppm and 5.5 ppm in a rat kidney during the period of the CT experiment. (A) The calibration curve of CT value versus iodine concentration in phantoms; (B) the dynamic changes of iodixanol and iobitridol concentrations in rat kidney after successful injection of two agents; (C) the calculated ratio between two nonequivalent amide protons. CT, computed tomography; CEST, chemical exchange saturation transfer.